



"Cutting Edge Solid/Hazardous Waste Management Research at FAU"

> Combining Science and Technology To Protect and Conserve Our Environment



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### http://labees.civil.fau.edu





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#### **Research Focus Areas**





#### **Urban Landfills** Do you think we might have a nuisance odor issue here?







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### State of the Art

#### **Nasal Ranger**

#### Olfactometry





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### **The Key Problem**



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#### • Subjectivity

Different individual perceptions of odors



### Challenges

- Ability to deal with adverse environmental or meteorological conditions (temperature, humidity, etc.)
- Source identification/competing odors
- Sensitivity at very low (<ppb) levels</li>
- Selectivity of odorants in real mixtures
- Synergistic and masking effects
- Subjectivity and individual perception





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#### **Developed Recommendations**



#### ODOR COMPLAINT LOG FORM FOR LANDFILL ODORS

#### **General Information**

- Date of odor complaint:
- Name of the person:
- Address of odor complaint:
- Time of odor complaint:
- Time AM/PM:
- Day in a week:
- Landfill in proximity to odor observation:

#### Description of Odor

What time was odor first detected?

What time was odor last detected?

Duration of odor (minutes or hours):

Location: indoors/outdoors:

Strength (1-5, with 1 being very light and 5 being very strong):

Character (type):

Meteorological Conditions at Time of Odor Observation

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Temperature (°F):

Wind speed (mph):

Wind direction:

Sky conditions:

Precipitation accumulation (inch):

Pressure (inch):

Humidity (%):



- Temperature/humidity
- Precipitation
- Atmospheric stability class (A – F)
- Pressure drop over the previous 24 hours

#### **Critical**

Odor complaints are expected imminently

#### **Severe**

Odor complaints are highly likely

#### **Substantial**

Odor complaints are a strong possibility

#### **Moderate**

Odor complaints are possible, but not likely

#### Low

Odor complaints are unlikely

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## **Potential Solution for Quantifying Odors**

- What if instead, we looked at this problem from the other direction?
  - For example, in wastewater there are simply too many organic constituents to distinguish
  - So we use an indirect, composite test for aggregate organics (i.e. BOD)
  - So why not do the same thing with odorants?
    - Since people only complain when odors are bad, is there a way to quantify an aggregate odor intensity (objectively)?



### **Used Biology as an Inspiration**





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### **Proposed Approach**

- Known protein isolates (hOBP2A) can bind with odorants in the µM-range
- Odor intensity is based on the number of bound receptors (non-specific?)
- If so, we can spectroscopically tag hOBP2A and its binding response will be concentration-dependent and follow Beer's Law (quantitative)
- May be reversible too



#### Mechanism



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### Objectives

- Create prototype biosensor
  - hOBP2A + 1-AMA tag complex
- Expose biosensor to odorants for various times
  - Single odorants and mixtures
- Measure spectroscopic signal to determine concentration dependence and quantitation range





### Quantitation Range for H<sub>2</sub>S

- Initial experiments showed maximum detection limit occurs around 4 minutes
  - 100 mL of biosensor @ 0.5 lpm
- Under these conditions, the maximum quantitation limit appears to be ~4-5 µg H<sub>2</sub>S





### Quantitation Range for H<sub>2</sub>S

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- Miniaturized to 10 mL, the limit is reached in 60 sec





#### **Other Gases Quantitation Ranges**

#### Peak intensity for CH<sub>4</sub> 0.5 slpm Protein:fluorophore 1:1.4



#### Peak intensity for NH<sub>3</sub> 0.5 slpm Protein:fluorophore 1:1.2





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### **Future Work**

- Analyze difference in results for acidic, neutral, and basic odorants
  - Initial slopes were same order of magnitude with subtle differences, suggesting we might be able to tease out different odorants in mixtures using binding affinities
- Experiments with mixtures
  - Standard landfill gas" (H<sub>2</sub>S, CH<sub>4</sub>, CO, N<sub>2</sub>)
  - Concentration range for detection/quantitation
  - Flow rate and reversibility









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#### **Funding Partners**





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## **Biogeochemical Clogging**

- Solids analyzed with x-ray diffraction
- Mixtures of leachate modeled with PHREEQC
- Clogging controls tested
  - Electronic pulsed scaling control
  - Dilution water
  - Acid injection
- Impacts to deep injection disposal well analyzed





### **Main Findings**

- Solids are mostly calcite (CaCO<sub>3</sub>) with mineral salt impurities and biological residues
- Practical solutions were implemented in the field for managing clogging impacts
- A new mechanism for clog formation was tested and duplicated in the lab
- This is leading to new ideas for clogging control in leachate collection system piping networks





## **Deep Injection Well Biofilms**

- Collect samples from 3 zones inside a deep injection well
  - Mineralogical analysis
    - Dominated by CaCO<sub>3</sub> (>90%)
    - Mineral salts (<10%)
    - High precipitation potential
  - Microbiological analysis
    - Methanogenic archaea & sulfate reducers
    - Entamoeba
    - Enterovirus and Pseudomonas phage





X-ray

Diffraction

**DNA** Analysis

### **Electrochemical Oxidation of Leachate**

- To determine the treatment performance of EOx, coupled with different pretreatment to remove COD and ammonia from landfill leachate
- Assess the generation of halogenated byproducts (THMs and HAA5)





#### **Raw Leachate**



Parameter	Range
рН	7.2 – 8.1
COD	$5,500 - 10,000$ mg/L as $O_2$
Ammonium-N	$2200 - 3200 \text{ mg/L} \text{ as } \text{NH}_4^+\text{-N}$
BOD	$500 - 700 \text{ mg/L} \text{ as } \text{O}_2$
Turbidity	230 – 1000+ NTU
Conductivity	30 – 80 mS/cm
Temperature	~25°C
TDS	23 – 54 g/L
Chloride	8 –14 g/L as Cl <sup>-</sup>



#### **EOx Reactors**

#### **OriginClear Roughing EOx Reactor**

#### **Magneli EOx Polishing Reactor**



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### **Main Findings**

- Pre-Treatment
  - Ozone
  - Fenton's Reagent
  - Lime
  - Two-stage EOx

- Higher chlorides increase removal of COD
- Longer reaction times also increase performance
- Maximum performance achieved from preozonation with Magneli EOx
  - COD removal = 52%
  - Ammonium-N removal = 52%
  - THMs = 5500 μg/L
  - HAA5 = 51,000 μg/L



# Energy from Waste





- In Florida, organic wastes make up 6-20% of the MSW
  - Only 2-5% is diverted from landfills
  - So ~2 million tons/yr landfilled
  - High moisture content (70%)
  - Low heating value (<2500 BTU/lb)\*</p>
- So, this stuff is undesirable for WTE, but ideal for anaerobic digestion

\*Compared to MSW (~5000 BTU/lb)

### **Anaerobic Digestion**

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- There are 1497 anaerobic digesters in the US
- 83% are used strictly for wastewater applications
- Recent innovations in co-digestion have unlocked the potential for cleaner biogas (65-75% CH<sub>4</sub>)
- If 10% of the digester feed is food waste, the biogas output can triple

### **Food Waste Anaerobic Digestion**

- Can we take advantage of unused digester capacity in the wastewater sector to generate more clean biogas from diverted food waste?
  - Ultimate digestibility testing
    - Biogas production
    - Methane content
    - VS destruction
  - Meat trimmings from butchers and FOG from grease traps
  - Various feedstock ratios
    - (1:2 1:10) at SRT = 7 28 days



### **Major Findings**

- Typical biogas production rate (0.2 0.5 m<sup>3</sup>/kgVS) for anaerobic digesters was surpassed at SRT > 21 days
- Maximum biogas production
  - Meat 1:7 (0.23 m<sup>3</sup>/kgVS)
  - FOG 1:4 (0.52 m<sup>3</sup>/kgVS)
- Methane production was >60% for both feedstocks
- With these values, diverting 0.6% of Palm Beach County food waste to one digester would produce enough power for 130 – 360 homes



# Watershed Management

- Catalog and analyze Florida watershed data and maps
- Develop a screening tool for flood management and apply to coastal and inland watersheds
- Develop guidance for preparing watershed management plans to improve CRS score





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